

CF₄/H₂ Plasma Cleaning of Graphene Regenerates Electronic Properties of the Pristine Material

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ABSTRACT

The impact on the electronic and structural properties of CVD graphene transferred onto SiO₂/Si of continuous H₂-based plasmas, used to remove sticky residues composed of PMMA and Si-based nanoparticles at the surface, was investigated. By combining X-ray photoelectron spectroscopy (XPS) and Atomic Force Microscopy (AFM), we found that H₂ plasma treatment, which allows simultaneous etching of Si-based nanoparticles and PMMA, causes fragmentation of CVD graphene layer into nanoplatelets and subsequent etching of the uncovered SiO₂/Si surface. We added CF₄ into the H₂ plasma to allow the selective etching of Si-based impurities while maintaining the quality and stability of the CVD graphene. The increase in sp²/sp³ ratio and decrease in Si-C bonds, evaluated from XPS analysis, reveals the removal of all residual contamination. AFM analysis confirms the efficient and selective etching of residues from the surface of graphene, which displays a microscopic corrugation due to the weak coupling with the SiO₂/Si substrate. The established CF₄/H₂ plasma processing generates, however, cracks along grain boundaries in CVD graphene, which is responsible for unusual transport properties. Characterization of local chemical structures using Raman spectroscopy reveals that CVD graphene layer is essentially undamaged under the CF₄/H₂ plasma and the dehydrogenation is incomplete in the subsequent annealing at 400 °C. The local electronic structure is probed using reciprocal-space photoemission electron microscopy (k-PEEM) and reveals a small, negative shift below 0.1 eV of the Dirac point with respect to Fermi level, which is consistent with n-doping caused by trapped hydrogen species at the interface. Threshold photoemission electron microscopy (PEEM) analysis establishes the work function of CVD graphene to be 4.57 eV. This study reveals that the optimized cleaning process almost recovers the original properties of quasi-freestanding graphene.

Keywords: H₂-based plasma treatment; silicon and PMMA contamination; graphene cleaning; hydrogen intercalation; electronic and structural properties; photoelectron spectroscopy; Raman spectroscopy; photoemission electron microscopy.

I. Introduction

The emerging need for high-speed electronics and low energy consumption devices has motivated researchers to explore the potential applications of graphene and 2D materials due to their outstanding properties.¹⁻⁴ Furthermore, the controlled modification of their properties has a potential to complement or replace current materials toward task-specific application in catalysis, energy, and sensing.^{5,6} These have led to extensive research efforts to synthesize large area, high-quality monolayer graphene.⁷ For instance, the growth of graphene on various transition metal substrates via chemical vapor deposition (CVD) route is investigated. Much progress has been made for rigorous control over shape, size, and the uniformity of domains in polycrystalline graphene.^{8,9} Recently, a millimeter to centimeter-sized graphene single crystals was achieved by modulating the nucleation site density on Cu substrates.¹⁰⁻¹² Moreover, tailoring the physical properties of CVD graphene for device applications requires more attention because it is sensitive to the influence of their immediate environment, including the supporting substrate, defects and surface contamination arising from the typical manufacturing process for fabricating 2D material devices.¹³⁻¹⁵ The most known contaminants include poly (methylmethacrylate) (PMMA), used as a mechanical support to transfer CVD-grown graphene onto a target substrate or as a photolithographic mask for patterning, and Si-based nanoparticles impurities resulting from the CVD oven and the used-masks.^{14,16,17}

A significant effort has been devoted to clean graphene while maintaining its high crystallinity and film quality.¹⁶⁻²³ Among such methods, plasma processing is considered as one of the most practical approaches for large-scale graphene treatment.^{22,24,25} We recently reported the detailed mechanism of the H₂ plasma selective etching of sacrificial PMMA film on CVD-grown graphene on Cu.¹⁷ In order to prevent graphene damage, the etching process has been accomplished by using plasma conditions in which the surface is bombarded by light and low energy ions (10 eV H_x⁺ ions)¹⁷. Even if the H₂ plasma is shown to be efficient in removing the two types of residual PMMA, i.e. the amorphous PMMA^A and the self-organized 2D structures PMMA^G, two reasons make it inconvenient to use. The H₂ plasma cleaning process leads to mobility and reorganization of Si-based nanoparticles impurities, which has been shown to slow down the lateral etching of PMMA^G, because Si atoms tend to accumulate at the edges of PMMA^G flakes thus inhibiting their lateral etching by hydrogen radicals.¹⁷ Furthermore, for high-performance device realization, it has been reported that H₂ plasma processing of graphene transferred onto the SiO₂ substrate may lead to the lift-off of the graphene layer.¹⁶ This is due to the penetration of protons through graphene and their recombination that forms an H₂ gas between SiO₂ and graphene.¹⁶ Meanwhile, it is shown that conditioning and cleaning the Al₂O₃ plasma reactor walls by a short SF₆ plasma treatment can remarkably improve the reproducibility of the surface processing by forming a stable AlF₃ thin layer on the reactor walls.²⁶ In fact, this plasma processing step, prior to any surface treatment, is required to prevent the release of highly reactive parasitic species such as O, which may affect the quality of graphene during processing.²⁶

Here we present the impact of CF₄ addition in the H₂ inductively coupled plasma (ICP) cleaning processes, with a focus on the structural and electronic properties of polycrystalline CVD graphene film with few micron-size grains transferred onto SiO₂/Si substrate. Indeed, fluorocarbon plasmas have long

been known to be efficient for etching Si-based materials²⁷ and the motivation of adding CF_4 to H_2 is to clean simultaneously carbon-based and silicon-based residues from the graphene surface. To this end, we performed a comparative study of graphene surface modification during etching with H_2 and CF_4/H_2 continuous ICP plasmas. The cleaning efficiency and the properties of the plasma-treated CVD graphene are investigated using XPS, AFM, and Raman in conjunction with PEEM/ kPEEM. The impact of the plasma treatment on long-range electrical transport through CVD graphene is also discussed.

II. Cleaning of CVD graphene surfaces by H_2 -based plasma: choice of plasma conditions

A simplified schematic that illustrates the transfer process at the origin of PMMA contamination as well as the cleaning process is presented in Figure 1. Basically, during PMMA solution-spinning, a uniform solid film of PMMA was formed after the evaporation of a solvent. Thermal treatment was employed to improve the adhesion of PMMA to graphene, which results in the reorganization of PMMA networks at the interface. Previous studies have shown three types of PMMA¹⁷; (1) thick layer of PMMA^B associated to the disordered structure, (2) thin layer of PMMA^A associated to disordered and dense structure at the interface and (3) thin layer of PMMA^G associated to molecules, including PMMA and adsorbed solvent molecules, organized into 2D layers on graphene. Usually, after the transfer is completed, acetone is used to enable the etching of PMMA^B, while the plasma process is used for etching PMMA^{A+G} and Si-based materials.

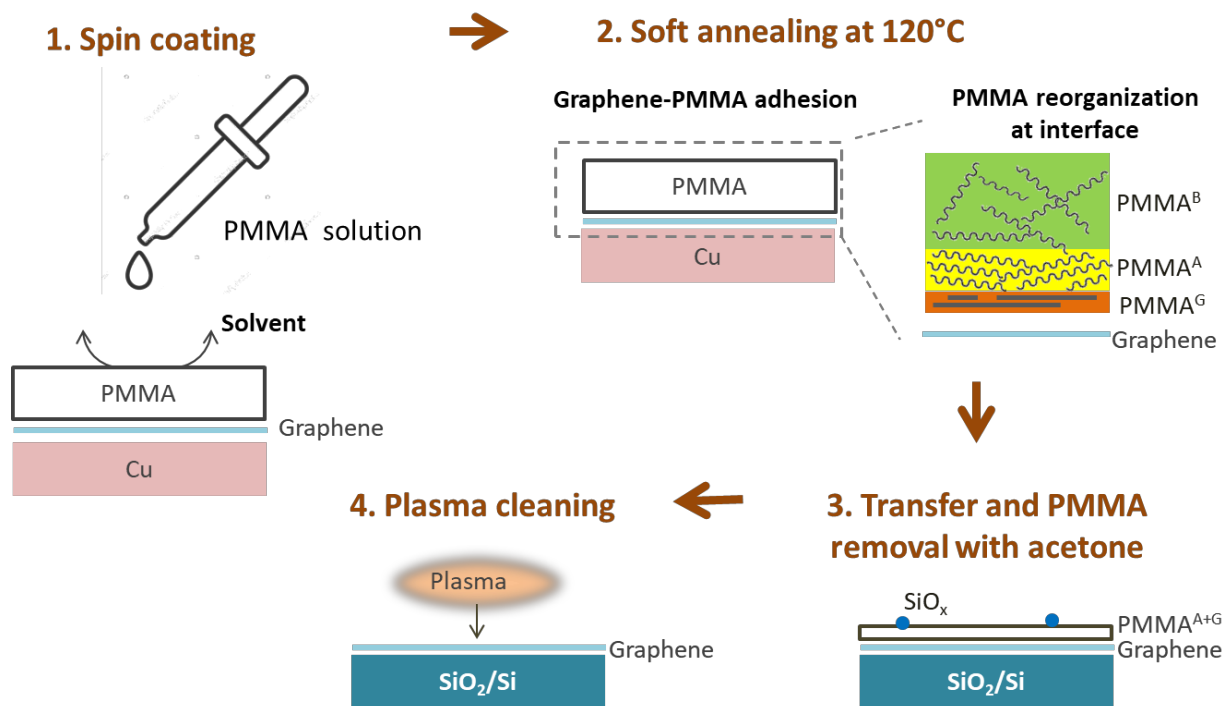


Figure 1: Schematic illustration of the CVD graphene transfer and cleaning processes.

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3 To recover the electrical properties of the quasi-free-standing graphene layer, displaying zero band gaps
4 and a band crossing at the Fermi level, an establishment of optimal plasma processing conditions is
5 paramount. For this purpose, we have investigated the effect of three relevant plasma conditions in H₂
6 and CF₄/H₂ chemistry. The main challenge consists in eliminating the Si-based nanoparticles, which
7 usually resists to hydrogen radicals and diffuses on graphene until they stick on edges of the PMMA^G
8 residues thus slowing down their elimination¹⁷.
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11 Typically, independent control of ion energy and plasma density can be provided by adjusting the source
12 power. Thus, to achieve uniform high-density plasma and low-energy ions that ensure the appropriate
13 surface cleaning of graphene, an intermediate ICP power of 800 W has been used. Here, we first report
14 the effect of pure H₂ plasma. In order to achieve the reactive etching of Si-based material over
15 graphene, we performed a comparative study between etching under conditions which are
16 predominantly ion-assisted or radical-assisted. This is realized by varying the chamber pressure from 40
17 mT (condition 1) to 200 mTorr (condition 2), respectively. When the pressure is raised, the H atoms flux
18 increases, while the H⁺ ions fluxes and energy are significantly reduced. Therefore these conditions were
19 chosen to analyze the respective roles of H atoms and ions on Si nanoparticles etching and graphene
20 damages.
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25 We also investigated the impact of CF₄/H₂ mixture plasma at high pressure, in which radical etching
26 predominates over ion enhanced etching (condition 3). The idea here is simply that F atoms are known
27 to be much more efficient than H to etch Silicon. The strategy is a two-step process: (i) a short step (10
28 s) in 0.5 % CF₄/H₂ plasma at 200 mTorr, used first to etch the silicon contamination, followed by (ii) a
29 long step (80 s) in pure H₂ plasma at 200 mTorr to remove the remaining carbon-based contamination
30 (PMMA^{A+G}). The optimal parameters, including CF₄ concentration and plasma exposure time, results
31 from systematic XPS and optical microscopy investigations of the surface modification under several
32 operating conditions. Our approach is based on the identification of conditions that allow the etching of
33 Si-based material without damaging graphene by maintaining the exposure time to the CF₄/H₂ plasma
34 shorter at lower CF₄ concentration that leads to (i) stable plasma and (ii) weak fluorine species diffusion
35 through PMMA toward the graphene layer. Indeed, the chemisorption of fluorine results in strong out-
36 of-plane corrugation of the graphene, which can make the structure more vulnerable to the hydrogen
37 radicals^{13,28}. It has been noticed that at higher CF₄ concentrations and longer exposure times, the plasma
38 process generates defects in the graphene.
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44 **III. Effect of surface plasma treatment on the chemical composition** 45 **and morphological properties** 46 47 48

49 To assess the change in surface chemical composition of the CVD graphene transferred onto 5 nm-
50 SiO₂/Si substrates after plasma etching of the residual contaminants, XPS investigations were
51 performed. Figure 2 displays the C1s core level XPS spectra of (i) pre-treated CVD graphene, (ii) after H₂
52 plasma treatment for 20 s, 80 s and 140 s using 800 W source power at a working pressure of 40 mTorr
53 (condition 1) and 200 mTorr (condition 2) and (iii) after CF₄-H₂ plasma treatment (condition 3). The
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3 removal of chemisorbed H, which behaves like sp^3 defect (C-H bonds) in graphene lattice, was carried
4 out via annealing in UHV at 400 °C. All spectra were normalized to Si2p component from the substrate.
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7 Figure 2. a displays C1s spectrum from pre-treated CVD graphene/SiO₂/Si, which consists of four
8 components from residual PMMA^{A+G} located at 285.7 eV (C-H), 286.4 eV (C-C), 287.0 eV (C-O), and 288.9
9 eV (O-C=O), and two carbon hybridized components; sp^2 (C=C) at 284.4 eV from PMMA^G and graphene
10 and sp^3 (C-C) at 285.0 eV from PMMA^G and defect structures in graphene.¹⁷ An additional C1s
11 component that is found at 283.9 eV is essentially attributed to the Si-C bonds present in the
12 contamination¹⁷. This component was used as an indicator to (1) identify the graphene-like structure of
13 PMMA^G layers, decorated at edge sites by residual Si atoms, and (2) monitor the variations in Si-C bonds
14 concentration arising from plasma interaction with Si-based nanoparticles enabling their diffusion or
15 their etching. In this study, we do not present the Si2p spectra because contamination components
16 overlap with those of the SiO₂/Si substrate. For line shape analysis, Voigt line shapes were used to fit
17 components, whereas modified Doniach-Sunjic line shape (with $a = 0.068$ and $g = 170$ meV) was used to
18 fit sp^2 component with a Shirley background subtraction. The sp^2 peak displays full width at half
19 maximum (FWHM) of 0.71 eV which appears to be higher than what was of the high-quality CVD
20 graphene grown on Cu displaying FWHM of about 0.45 eV, as can be seen in Figures S1 in the Supporting
21 Information. Thus, the presence of PMMA^{A+G} residues, in addition to constraint-based wrinkles, which
22 arises from the transfer process of graphene²⁹, could be the origins for the increase of FWHM.
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28 Figure 2. b displays C1s spectra from CVD graphene treated by low pressure H₂ plasma (condition 1), in
29 which the representing surface is subjected to a high flux of relatively low energy (about 10 eV)
30 hydrogen ions. This ion bombardment is required to etch the Si-based nanoparticle contamination. We
31 first notice a decrease in the PMMA components intensity after 20 s of plasma treatment. At this stage,
32 we assume that the plasma did not interact with graphene because it is still covered by a thin layer of
33 PMMA^G. Although the reactive ion etching shows a decrease in PMMA^G and Si-C bonds when the
34 process duration is extended to 80 s, the graphene starts to be seriously damaged. This is evidenced by
35 the large decay of the overall C1s intensity by a factor of 5, an increase in relative contribution of the sp^3
36 and the formation of C-O groups, found at 287.2 eV. If the process is further prolonged to 140 s, the C1s
37 signal is now divided by 10, the sp^3/sp^2 intensity ratio increases dramatically (53 % contribution of sp^2)
38 and we notice, in addition to CO, the presence of C-OH groups, seen at a binding energy of 285.8 eV,
39 arising from oxygen atoms occupying the defects and edge sites in graphene. This indicates a severe
40 degradation of the graphene structure by the plasma: after 140 s only flakes of graphene decorated by H
41 and O remains. The exact mechanism whereby the graphene transferred onto the oxide substrate is
42 damaged is yet to be elucidated. The ion energy is already high enough for protons to pass through
43 graphene, but does not allow for the strong sp^2 bond breaking, as has been demonstrated previously for
44 CVD graphene on Cu¹⁷. Thus, it is possible to consider (i) the presence of O species, originating from the
45 reduction of the SiO₂ substrate underneath graphene, that contribute to graphene etching and/or (ii)
46 the partial *lift-off* of the graphene or the formation of H₂ bubbles which leads to graphene cracks, as
47 discussed in a recent paper²⁶. Therefore, even if this plasma is able to remove PMMA^G and related-
48 silicon contaminations (after 140 s plasma exposure, the Si-C components becomes almost invisible), it
49 cannot be use to clean graphene.
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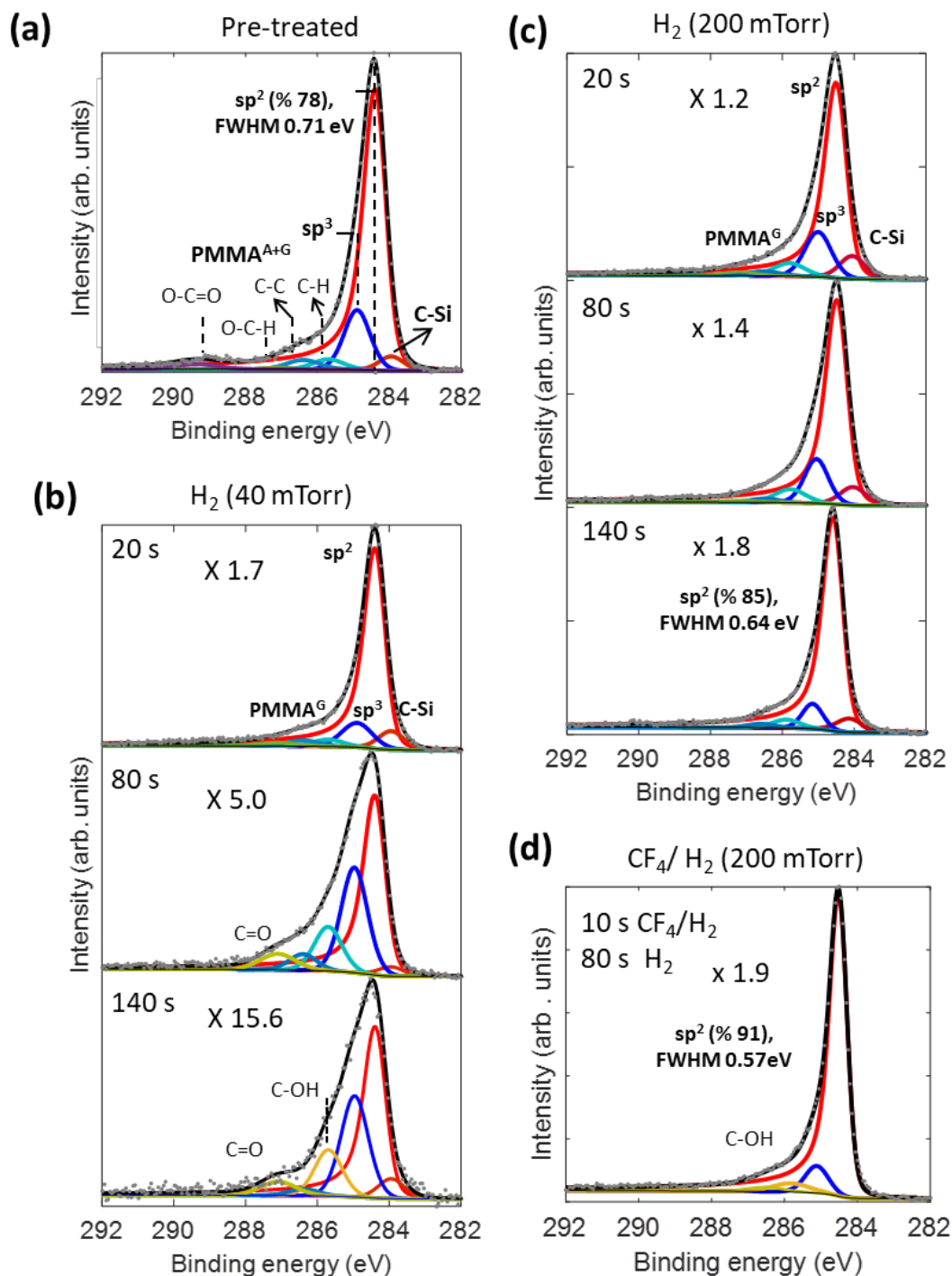


Figure 2: C1s core level spectra recorded before and after H₂-based plasma treatment of the CVD graphene transferred onto 5 nm-SiO₂/Si substrates at an ICP power of 800W: (a) pre-treated, (b) H₂ plasma at 40 mTorr for exposure time of 20s, 80s and 140s (c) H₂ plasma at 200 mTorr for exposure time of 20s, 80s and 140s and (d) CF₄/H₂ plasma at 200 mTorr.

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3 By contrast, the C1s spectra, shown in Figure 2. c, report the surface evolution under H₂ plasma working
4 at condition 2, related mainly to the hydrogen radical etching. A statistical approach of the plasma
5 phenomena suggests that reducing the ion-energy flux should significantly reduce the probability of
6 graphene degradation. The relative intensity variation reveals a decrease in the PMMA, Si-C, and sp³
7 components contribution as a function of plasma duration. The shape of C1s spectra remains unchanged
8 after 140 s while the Si-C component is still present indicating incomplete cleaning of Si-related
9 contamination, namely PMMA^G and Si-based nanoparticles. This result indicates that CVD graphene
10 processed with hydrogen radicals, did not suffer from damage and showed a higher intensity
11 contribution of sp² (85 %) with a full width at half maximum (FWHM) of about 0.64 eV. The assessed
12 FWHM of the sp² component is consistent with the presence of residual PMMA^G, which cannot be fully
13 eliminated by H radicals alone in the presence of parasitic silicon residues.
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18 Finally, Figure 2. d presents C1s spectrum obtained after plasma etching of PMMA^G and Si using CF₄ and
19 H₂ gases under condition 3. The deconvolution analysis indicates an sp² intensity contribution of 91 %,
20 which is higher than those previously reported, with a full width at half maximum (FWHM) of about 0.54
21 eV, close to result obtained on as prepared CVD graphene layer on Cu foil. In addition, the Si-C
22 component is not observed indicating the effectiveness of the plasma process to remove both the
23 PMMA^G and Si contamination. At this stage of the investigation, the exact origin of remnant sp³ and C-
24 OH components, associated to random defects in graphene, is unknown. To complement these
25 observations, the surface morphology changes with aforementioned hydrogen-based plasma
26 treatments are now examined.
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30 Figure 3 shows AFM images (5 x 5 μm²) and the corresponding surface roughness profiles, taken along
31 the indicated lines, of aforementioned CVD graphene after the final plasma treatment under the
32 condition 1, condition 2, and condition 3. AM-KPFM (Kelvin Probe Force Microscopy) images of
33 untreated CVD graphene transferred onto 5 nm-SiO₂/Si showing PMMA and Si-based nanoparticle
34 residual contamination are presented in Supporting Information S2. From all three of the
35 aforementioned conditions, it is assumed that only PMMA^G and Si-based contamination may be left on
36 the surface. Identifying the remaining Si-based contamination with AFM and KPFM is challenging
37 because the initial morphology of nanoparticles is not maintained under H₂ plasma (they diffuse on the
38 surface)¹⁷; while the remaining 2D layers of PMMA^G are expected to show an inter-plane distance similar
39 to that of graphite. Figure 3. a displays the surface topography, obtained after plasma treatment under
40 condition 1, and shows very rough structure, similar to the assembly of graphene nanoplatelets with an
41 average diameter of 6 nm, which is well evidenced in the enlarged AFM image (1 x 1 μm²). Although an
42 efficient removal of PMMA^G is achieved, this result shows that the graphene layer is fragmented by the
43 plasma, caused by the creation of the defects in graphene and its subsequent lateral etching, in good
44 agreement with previous XPS analysis. We also noticed a formation of 2 nm deep pores, which is
45 associated to plasma etching of uncovered SiO₂/Si substrate.
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52 The surface topography, obtained after plasma treatment under condition 2, is shown in Figure 3. b. This
53 indicates a fairly homogeneous surface. The measured roughness corresponds to the 1-2 interlayer
54 distance in graphite of 0.35 nm which can be explained by the heterogeneous coverage of graphene
55 surface with 2D-like PMMA^G layers of hundred-nanometer diameter, having an organizational structure
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similar to that of graphene, as reported in previous paper¹⁷. This result indicates that the complete etching of PMMA^G is not possible using H₂ plasma that does not damage graphene, which is in good agreement with previous statement derived from XPS results.

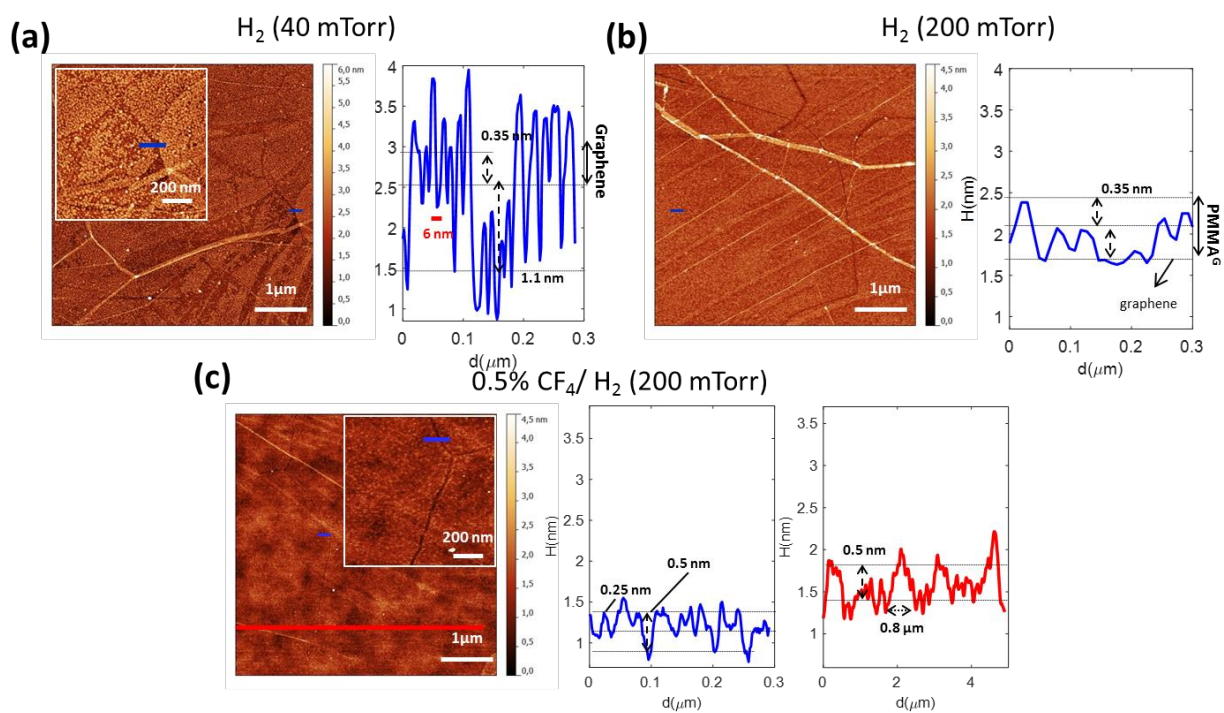


Figure 3: AFM images and the corresponding height profiles, taken along the indicated lines, of CVD graphene transferred onto 5 nm-SiO₂/Si substrate after plasma cleaning process (a) H₂ plasma at 40 mTorr (b) H₂ plasma at 200 mTorr and (c) optimized 0.5% CF₄/ H₂ plasma.

The surface topography, obtained after plasma treatment under condition 3, is seen in Figure 3. c. A nearly flat surface is observed. A short range roughness with an average amplitude of 0.25 nm (blue line profile), significantly lower than the interatomic distance in graphite, is observed. Since the XPS study indicates a good chemical structure of treated CVD graphene, the possible sources of this corrugation could be (i) a rough SiO₂/Si substrate surface²³, (ii) hydrogen passivation of SiO₂ surface and (iii) trapped hydrogen-based species at the surface that not completely desorb with thermal annealing treatment at 400 °C. We also identified the grains in CVD graphene layer having an average size of several microns, dissociating from each other at the grain boundaries. AFM topography image at a smaller field of view (1 x 1 μm²) shows a maximum distance between two gains of 25 nm. Thus, the important height change seen is attributed to the plasma etching along grain boundaries of CVD graphene (strongly disordered area and particularly vulnerable to plasma processes) followed by the SiO₂/Si substrate etching. Therefore, the origin of small contribution of sp³ and OH components, seen in C1s, is attributed to grain boundary/edge defects and plausible trap species at the interface. A long-range undulation (red line profile) with 0.5 nm out-of-plane deformation is clearly observed and associated to ripples in graphene with nearly quasi-free-standing behavior^{30,31} due to low interaction with SiO₂/Si substrate. This may also explain the reduction of wrinkles from substrate constraint. Usually, the AFM analysis at this stage of

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3 surface cleaning is difficult to obtain because of electrostatic interactions of graphene with the tip and
4 weak coupling of graphene layer with the SiO₂/Si substrate. Basically, these results indicate that the
5 surface cleaning of single crystal graphene, a sp²-bonded carbon area, is effective using a mixture of
6 CF₄/H₂ while maintaining its quality, in contrast to processing based on pure H₂ plasma, where the
7 PMMA^G layers resist to etching.
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10 Although this treatment is efficient on single crystal graphene, the discontinuity at the grain boundaries
11 in CVD graphene layer leads to unusual electrical transport measurements, as reported in Supporting
12 Information S3. Indeed, the measurement of the resistivity as a function of the back gate voltage (V_{BG})
13 shows a charge neutrality point above 50 V and an increase in resistivity suggesting that treated-CVD
14 graphene layer is heavily hole-doped and behaves like an insulator. In order to assess the effect of the
15 established CF₄/H₂ plasma treatment on the local structural and electronic properties of plasma-treated
16 CVD graphene, Raman and PEEM analysis are realized and described in the next section.
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20 **IV. Structural and electronic properties of optimized plasma-treated** 21 **CVD graphene: defect and doping** 22 23 24

25 To probe the impact of optimized-plasma processing on the structure of CVD graphene, Raman analysis
26 has been performed using a laser at an excitation wavelength of 532 nm. Here, 300 nm-SiO₂/Si
27 substrates are used to support CVD graphene because Raman signals, obtained from CVD graphene
28 transferred onto 5 nm-SiO₂/Si substrates, are essentially weak. Previous XPS and AFM studies have
29 addressed the plasma-induced irreversible defects in graphene; the systematic annealing after plasma
30 treatment was carried out to desorb hydrogen from the surface and thus prevent a complication in the
31 interpretation of the results. In this section, the surface characterization by Raman spectroscopy is
32 reported before and after annealing in order to assess (i) the localization of adsorbed hydrogen and (ii)
33 the doping level in graphene. Figure 4 shows four Raman spectra, consisting on the average
34 measurements taken at a different position, of CVD graphene (i) without any treatment, (ii) after
35 optimized plasma treatment and (iii) after optimized plasma and thermal annealing treatment.
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40 Raman spectrum of untreated CVD graphene, shown in Figure 4. a, displays two typical peaks, G band ~
41 1592 cm⁻¹ and 2D band ~2695 cm⁻¹, involving phonons near the K-point at the Brillouin zone. The 2D
42 mode decomposition in one component and G and 2D bands bandwidths of 13 and 35 cm⁻¹,
43 respectively, indicate that analyzed areas are typically monolayer graphene. The defect-induced D band
44 ~ 1354 cm⁻¹, arising when the translational symmetry of graphene is broken, is barely seen; a trivial
45 contribution is mainly given by defects at grain boundaries in the CVD graphene¹³. Figure 4. b displays
46 two spectra acquired at different positions, and obtained after plasma processing; these exhibit a
47 different intensity of defect peaks. The D band intensity dependence with position indicates
48 inhomogeneous distribution from plasma processing with localized areas displaying significant defect
49 density. The intensity ratio of D and G peaks (A_D/A_G), about 0.6 to 1, suggests low to moderate defect
50 density¹³. Another defect band, D', merge at 1620 cm⁻¹, indicating phonon confinement effect on the sp²
51 region³²⁻³⁴. The appearance of this band reveals the formation of structural disorder that has an average
52 defect separation on the order of a few nanometers.
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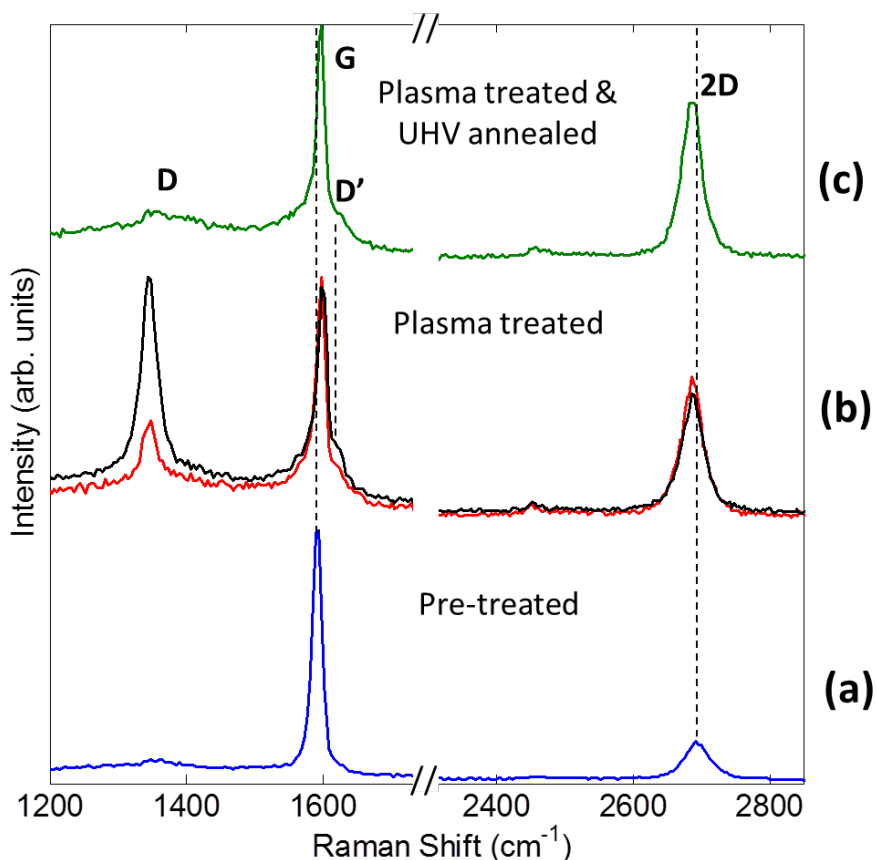


Figure 4: Raman spectra of CVD graphene transferred onto 300 nm-SiO₂/Si before and after surface processing. (a) Pre-treated, (b) plasma treated and (c) plasma treated and annealed.

Plasma-induced defect peaks may merge from irreversible vacancy defects or reversible sp^3 -defects, induced by hydrogenation, through thermal annealing.^{13,33} Based on literature^{35–38}, the variation in G and 2D peaks frequency depend on doping type; unlike the blue shift in the G peak, the blue-shift with p-doping and the red-shift with n-doping are reported in 2D peak. After plasma processing, we notice that G peak was shifted toward higher wavenumber by 6 cm^{-1} (blue shift), whereas 2D peak was shifted toward lower wavenumber by 8 cm^{-1} (red shift) revealing n-doping effect. The impact of residual PMMA-induced p-doping, acting as an electron acceptor, on the peak positions of the G and 2D have been studied, which reveals a weak blue-shifted of 1-4 cm^{-1} order of magnitude^{39,40}. Thus, the observed n-doping can be explained by effective removal of PMMA residue and hydrogen chemisorption-induced n-doping through electron delocalization as result of C=C bonds breaking and C-H bonds formation^{33,41,42}. An increase in the intensity ratio of 2D to G peaks (A_{2D}/A_G), which depends on the carrier density in graphene³⁷, is revealed. This observation confirms that the PMMA residues inducing p-doping are effectively removed. The annealing at 400 °C, performed to recover the undoped graphene surface, seems to boost more the intensity ratio A_{2D}/A_G , indicating that hydrogen adsorbates on graphene are removed, as reported in Figure 4. c. Although a considerable decrease in D intensity peak with annealing, associated to suppression of sp^3 -type defects in hydrogenated graphene^{33,34,43}, is shown, the

initial surface is not totally recovered. The remaining contribution is associated to trapped species at the interface and incomplete dehydrogenation which is supported by the unchanged 2D and G peaks position (n-doping). An annealing at a higher temperature may improve dehydrogenation processing but may also induce irreversible defects in graphene^{23,44}. Unusually, a weak background $\sim 1200\text{-}1600\text{ cm}^{-1}$ overlapping G and D' bands appears after plasma treatment which is consistent before and after annealing CVD graphene. Based on literature⁴⁴, this may be associated to undesorbed fragments of decomposed PMMA chains. These results are in agreement with the aforementioned XPS and AFM analysis, indicating that optimized plasma processing does not affect the graphene quality.

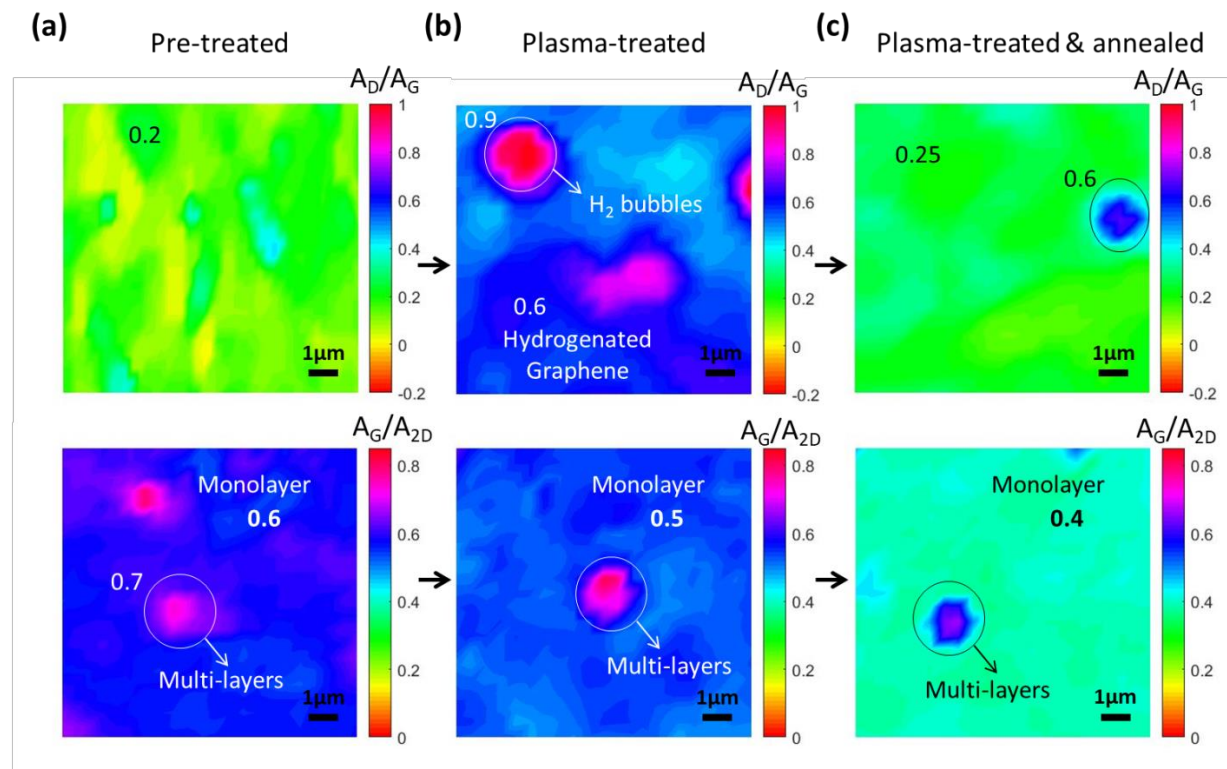


Figure 5: Intensity ratio maps of D to G band (A_D/A_G) and G to 2D band (A_G/A_{2D}) of CVD graphene transferred on 300 nm-SiO₂/Si: (a) as prepared, (b) plasma treated and (c) plasma treated and annealed.

In order to further demonstrate the effect of plasma and annealing treatment, intensity ratio maps of A_D/A_G and A_G/A_{2D} are reported in Figure 5. The measurements were taken from a ($10 \times 10\ \mu\text{m}^2$) area with a lateral resolution of $1\ \mu\text{m}^2$. Before treatment, A_D/A_G map, displayed in Figure 5 .a, basically reveals a weak and homogeneous spatial defect contribution. With A_G/A_{2D} map, seen in Figure 5 .a, we identify regions of multi-layer graphene, displaying higher intensity ratio of G to 2D bands. After plasma treatment, the A_D/A_G map, reported in Figure 5. b, shows an increase of defect intensity over the entire surface, displaying mainly two types of regions with moderate and low defect density. Most moderate defect density-regions have a circular shape of 1-2 μm diameter, suggesting the formation of hydrogen bubbles at the interface between the CVD graphene layer and the SiO₂/Si substrate. By observing A_D/A_G and A_G/A_{2D} maps, we found that hydrogenation coverage is independent on the number of layers. In principle, graphene exposed to H atoms only should not undergo a hydrogenation because it is

protected from chemisorption and penetration by the potential barrier originated from the π -electron cloud at sp^2 -hybridized crystal structure^{17,45}. However, hydrogenation happens because the CVD graphene surface is corrugated rather than flat, which significantly reduces the energy barrier. Furthermore, even if the ion energy and flux are significantly reduced at high pressure, the graphene remain bombarded by a low flux of H_x^+ ions with a high enough energy to chemisorb. After annealing, A_D/A_G and A_G/A_{2D} maps, seen in Figure 5. c, show values close to that of the reference sample, explained by a decrease in the total density of defects formed from hydrogenation, but not a complete disappearance (that may induce a slight n-doping). Basically, these results indicate that the optimized plasma-induced defect is a reversible process and can be recovered by annealing.

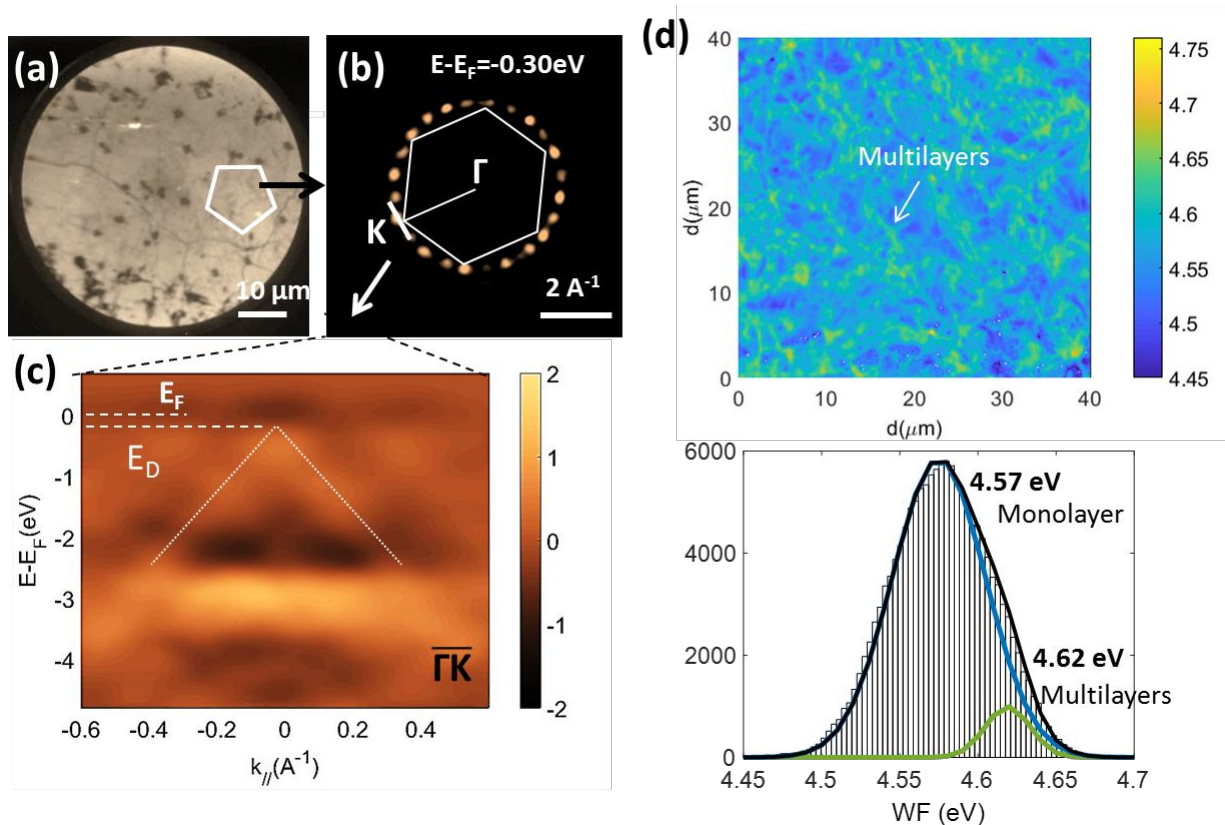


Figure 6: Plasma treated and annealed-CVD graphene transferred onto 5 nm- SiO_2/Si : (a) energy filtered PEEM image of 63 mm FoV acquired at a threshold energy ($E-E_F = 4.6\text{ eV}$), (b) energy filtered k-PEEM image of a 5.1 \AA^{-1} FoV acquired at $E-E_F = 0.3\text{ eV}$ from $10\text{ }\mu\text{m}^2$ selected area and (c) corresponding band structure along the $\overline{\Gamma K}$ direction showing a Dirac point shift with respect to Fermi level by 0.1 eV. (d) Work function map ($42 \times 42\text{ }\mu\text{m}^2$) derived from the threshold PEEM analysis with the corresponding histogram.

2D-band structure imaging of individual single-crystal domain in CVD graphene, transferred onto 5 nm- SiO_2/Si substrate, after surface treatment is performed using momentum-resolved photoemission electron microscopy (k-PEEM) in UHV. Figure 6. a shows an energy filtered PEEM image (direct space, 63 μm FoV) acquired near threshold ($E-E_F = 4.6\text{ eV}$), and showing mainly monolayer graphene with random

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3 small multilayer (dark contrast). Figure 6. b shows an energy filtered kPEEM image (5.1 \AA^{-1} FoV)
4 acquired close to the Fermi level ($E-E_F = 0.3 \text{ eV}$) from a 10 \mu m selected area, which displays only
5 monolayer graphene, shown in Figure 6. a. The obtained k-PEEM patterns show a set of six-fold
6 symmetry photoemission spots related to the Dirac cones from grains in the CVD graphene. These are
7 rotated relative to each other by approximately 30° , indicative of multiple domains rotated in-plane
8 within the selected region. The corresponding band structure along the $\bar{\Gamma}\bar{K}$ direction, characteristic of
9 one domain, is displayed in Figure 6. c. A negative energy shift of $\sim 0.1 \text{ eV}$ of Dirac point with respect to
10 the Fermi level is revealed, suggesting n-doping in agreement with Raman observations. The Dirac cone
11 is close to the Fermi level; the complete desorption of hydrogen by annealing may lead to zero intrinsic
12 doping and therefore the Fermi level at Dirac point.
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17 Figure 6. d shows the work function map ($42 \times 42 \text{ \mu m}^2$) and corresponding histogram, recorded on the
18 aforementioned sample. This map indicates a homogenous surface with an average work function value
19 of $4.57 \pm 0.2 \text{ eV}$ for graphene monolayer, which is very close to that evaluated for multilayers areas of
20 $4.62 \pm 0.2 \text{ eV}$. The assessed value is in good agreement with that established in literature for un-doped
21 single-layer graphene⁴⁶⁻⁴⁸, where the Fermi level falls in the Dirac point. However, some other studies
22 report a lower value of 4.2-4.3 eV for single layer graphene supported on SiC or SiO₂/Si substrates^{49,50},
23 which most likely forms Schottky junction at the interface with p-type charge transfer doping, or have
24 adsorbates at surface^{48,51}. Thus, the weak coupling between the graphene layer and SiO₂/Si substrate, as
25 suggested by AFM observations, and lack of surface doping, evidenced by a k-PEEM study showing a tiny
26 change in graphene Fermi level E_F , may explain the obtained work function value. Indeed, the shift of
27 the Fermi level due to doping-induced charge carrier density modulation mostly impacts the graphene
28 work function; it essentially increases with the electron-doping and decreases with the hole -doping⁵²⁻
29 ⁵⁴. A notable impact of residual trapped H species, seen to increase slightly charge carriers, is not
30 discerned, while moderate coverage of hydrogen can yield a larger work function change^{33,41,55}. The
31 presence of discontinuous coverage of PMMA^G on graphene induced-hole doping shows a small increase
32 in work function, evaluated to 4.65 eV. The corresponding work function map is included in the
33 Supporting Information S4. The recovering of expected graphene work function with optimized plasma
34 treatment, confirms the effectiveness of the process for removal of PMMA and Si contamination.
35 Furthermore, the low-pressure ions-based H₂ plasma treatment dramatically lowers the work function
36 from 4.65 eV, in presence of PMMA^G, down to 3.5 eV as a result of the introduction of vacancy-based
37 defects in graphene, as reported in Supporting Information S4. Therefore, modulation and tuning of H₂-
38 based plasma conditions is crucial for improving surface graphene cleaning.
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46 Conclusion

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48 The transfer process of CVD graphene grown on Cu foil to other substrates usually generates two types
49 of contamination: PMMA and Si nanoparticles. A cleaning method based on H₂ plasma treatment in
50 combination with annealing at 400 °C in UHV yields the removal of most of the residual polymer but is
51 not able to fully etch Si-based contamination, including PMMA^G layers, without damaging graphene
52 structure. The surface cleaning, obtained by CF₄/ H₂ radical etching, eventually results in the effective
53 removal of the Si-based nanoparticles and residual PMMA. Although plasma does not affect the
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3 properties of the sp^2 structure, at the grain boundary the formation of cracks of a few nanometers
4 length is observed because of typical weak binding (vulnerable region). This results in unusual electronic
5 properties displaying such heavily p-doped and resistive CVD graphene layer while the local electronic
6 structure investigation reveals graphene that is somewhat n-doped by residual adsorbed hydrogen,
7 displaying ~ 0.1 eV shift in the Dirac point. Although our investigation shows that the optimized plasma
8 treatment almost restores the original electronic properties of single crystal and un-doped graphene
9 layer, improvement in the cleaning process is needed. Ideal conditions are a high flux of thermal H and F
10 atoms without ions and at high temperatures (PMMA etching is thermally assisted). This could be
11 achieved in downstream plasma thus reducing the amount of hydrogen adsorbed and H trapped
12 underneath graphene, and perhaps avoid the grain boundaries deterioration in CVD graphene.
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16 17 **Experimental**

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19 **The plasma etching** is carried out in an industrial high-density ICP source (DPS AdvantEdge™) from
20 Applied Materials design to etch 300 mm diameter wafers⁵⁶. To prevent the presence of contamination
21 deposited during the etching process and parasitic O in the plasma due to Al_2O_3 reactor wall corrosion in
22 reactive H_2 plasma, the plasma reactor wall is cleaned and coated with a resistive AlF_3 layer using typical
23 industrial cleaning process with a SF_6/O_2 plasma⁵⁷ and conditioning NF_3 plasma in between each
24 experiment. For the same reason, we use 300 mm diameter wafer holders made of fluorinated Al_2O_3
25 ($AlF_3@Al_2O_3$ thin layer on a silicon wafer) to introduce and hold small graphene samples in the reactor.
26 The effect of plasma and reactor wall coating on the treatment of fragile 2D material has been detailed
27 on another paper²⁶. Samples are stuck with kapton™ tape on the wafer, and the temperature is kept at
28 $65^\circ C$ with a He backside cooling. The ICP in a continuous (CW) mode has been used to process the
29 graphene samples.
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34 **Samples** used in this study are commercially produced large area CVD graphene monolayer from
35 Graphenea Company. We use two different types of graphene samples: PMMA-assisted CVD graphene
36 transfer onto SiO_2 (300 nm thick)/Si and Si (with native SiO_2) substrates. The sacrificial thin layer of
37 PMMA (400 nm), used for transfer processing, is removed with acetone.
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40 The **XPS** measurements over an area of few mm^2 were carried out at a base pressure 10^{-9} mbar in a
41 Multiprobe spectrometer (ScientaOmicron) equipped with a monochromatized Al $K\alpha$ source (1486.6 eV)
42 and a 128 channel, parallel detection Argus electron analyzer. The overall energy resolution was 270
43 meV. The emission angle is fixed to 20° with respect to the surface normal. Thus, the probed depth is
44 about 8 nm for PMMA and graphene ($\lambda \sim 3$ nm). The spectrometer chamber is connected to a preparation
45 chamber in which samples could be annealed in ultrahigh vacuum (10^{-9} mbar) in order to remove
46 chemisorbed plasma species and impurities.
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50 The **X-PEEM and k-PEEM** experiments were performed in an ultra-high vacuum chamber using a
51 NanoESCA spectromicroscope (ScientaOmicron), described previously elsewhere⁵⁸. The
52 spectromicroscope is equipped with three different monochromatized sources, Al $K\alpha$ (1486.6 eV), He II
53 (21.2) and Hg (4.9 eV). A double pass hemispherical energy analyzer was used to compensate single
54 analyzer aberrations imaging. A $68 \mu m$ field of view was used with 12 kV extraction voltage and about
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3 1.8 mm sample-objective lens distance. The overall energy resolution of the analysis was 200 meV (Pass
4 Energy of 50 eV, slit 1). A contrast aperture of 500 μm was used corresponding to a lateral resolution of
5 about 0.5 μm . The energy filtered PEEM image series was acquired at the photoemission threshold
6 region using He II source and in the C1s, Si2p region using X-ray source. After correction for the Schottky
7 effect of 98 meV because of the high extractor field, the fit of the photoemission threshold spectra with
8 error function can be used to directly measure the local work function. Then, the series of images taken
9 at an increasing photoelectron kinetic energy enables us to determine work function map obtained from
10 the fit of the spectrum for each pixel. In the same way, the XPS spectra of a selected small area were
11 acquired. The k-PEEM results were acquired using He II source and fully opened aperture 1500 μm . This
12 was required to image a sufficient portion of reciprocal space in order to cover a full Brillouin zone. The
13 area of interest on the sample surface is chosen by a field aperture situated in an intermediate image
14 plane that was closed down to about 10 μm . A transfer lens then projected the 1500 μm diameter disk
15 of the focal plane via the energy analyzer onto the detector giving a 2D k-space dimension of about
16 $\pm 2.5^\circ \text{A}^{-1}$ around the point. The wave vector resolution is $\sim 0.05^\circ \text{A}^{-1}$. The detector response was
17 corrected by the flat field of the detector and camera defects were eliminated using dark images. With
18 our graphene/SiO₂/Si samples, it was found out that additional sample illumination with UV light (4.9
19 eV) using a Hg arc lamp allows for stable measurement.

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26 The **AFM** measurements were carried out in tapping mode on a Dimension Icon from Bruker. Standard
27 silicon tips from Budget Sensors (TAP300Al-G) with a first resonant frequency of 300 kHz range and a
28 nominal spring constant of 40 N/m were used. Areas of 1 x 1 μm^2 and 5 x 5 μm^2 with a 512 x 512 pixels
29 resolution were typically analyzed.

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32 The **μ -Raman** measurements were performed with a commercial Witec Alpha 500 spectrometer set-up
33 with a dual-axis X-Y piezo stage. Dual Laser with 532 nm excitation wavelength was used with 0.1 mW
34 power. The grating had 1,800 lines per mm in the best case, conferring a spectral resolution of 0.01 cm^{-1} .
35 Raman spectra were recorded in the air with a Nikon x100 objective focusing the light on a 320 nm-
36 diameter spot. Raman shifts are calibrated using the silicon line as a reference. Raman cartographies
37 recorded on graphene samples were computed with the Labspec5 program.

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40 **Electron transport properties** involved a Signal Recovery PAR Lock-in amplifier with a current bias
41 detection using a 10 Mega Ohm bias resistor in series with the sample. The silicon sample is used as a
42 back gate to measure the field effect of the graphene sheet.

43 44 45 **Acknowledgments**

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49 measurements were made on both the CEA Minatec Nanocharacterization Platform (PFNC) and the
50 clean rooms of CEA-LETI.

51 52 53 **Present Addresses**

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Author contributions

Gilles Cunge, Olivier Renault and Vincent Bouchiat were involved in planning and supervised the work and the findings. Gilles Cunge is specialized in the plasma-based material processing. Olivier Renault is specialized in the surface analysis and especially in the XPEEM/ KPEEM characterization technique. Vincent Bouchiat is specialized in synthesis of 2D material and their electrical characterization. Djawhar Ferrah and Gilles Cunge contributed to the implementation of the plasma processes. Djawhar Ferrah processed and subsequently characterized samples with XPS and PEEM spectroscopy helped by Olivier Renault. Nicolas Chevalier and Denis Mariolle performed the AFM measurements. Vincent Bouchiat and Daniil Marinov performed the Raman measurements and electrical measurements. Djawhar Ferrah processed the experimental data, performed the analysis, drafted the manuscript and designed the figures. Denis Rouchon helped in interpreting the Raman results and characterization (not reported here). Javier Arias-Zapata helped in sample preparation and SEM analysis (not reported here). All authors discussed the results and commented on the manuscript.

ASSOCIATED CONTENT

Supporting Information available: [C1s spectrum obtained from CVD graphene grown on Cu foil; AM-KPFM imaging of pre-treated CVD graphene transferred onto 5nm-SiO₂/Si; Electrical transport properties of optimized plasma-treated CVD graphene; Work function measurement on pure H₂ plasma-treated CVD graphene; Pressure tuning for plasma cleaning process optimization]

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